

**Robert E. Lutz and Charles J. Kibler.** The 1,2,4-Trimesityl 1,4-Diketones and Related Compounds, Including the Stereoisomeric Mono and Dienols.

Page 369. Column 1, sixth line from bottom should read, "A solution of the 4-monoenolate-A (II) prepared from 3 g. of the iododiketone and 12 cc. of . . ."

Page 370. Column 1, line 15, for "saturated 1,4-diketone III" read "unsaturated 1,4-diketone (I)."—R. E. LUTZ and C. J. KIBLER.

**J. H. Simons, S. Archer and D. I. Randall.** Hydrogen Fluoride as a Condensing Agent. X. Rearrangements.

Page 485. Column 1, second line from bottom, for "2-hydroxy-4-methyl-diphenyl sulfone" read "2-hydroxy-5-methyl-diphenyl sulfone."

Page 486. Column 1, line 34. For "2-methoxy-4-methyl-diphenyl sulfone" read "2-methoxy-5-methyl-diphenyl sulfone." Column 2, line 9 and bottom line for "2-hydroxy-4-methyl-diphenyl sulfone" read "2-hydroxy-5-methyl-diphenyl sulfone."—J. H. SIMONS and AARON ADDELSTON.

**G. H. Messerly and J. G. Aston.** The Heat Capacity and Entropy, Heats of Fusion and Vaporization and the Vapor Pressure of Methyl Chloride.

Page 887. Table II, 2nd line from bottom of the column headed *T* obsd., °K. for "248.943" read "248.192."—G. H. MESSERLY.

**Louis F. Fieser and Lloyd M. Joshel.** 9-Methyl-3,4-benzfluorene.

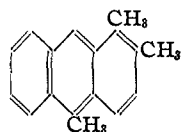
Page 957. In formula III the subscript 2 should be transferred from C to H.—LOUIS F. FIESER.

**Melvin S. Newman and Lloyd M. Joshel.** The Synthesis of 2-Methyl-3,4-benzphenanthrene.

Page 973. Column 2, line 35: for "picrate" read "sym-trinitrobenzene derivative."—M. S. NEWMAN.

**Louis F. Fieser and Thomas G. Webber.** *meso*-Alkyl Anthracenes.

Page 1360. Formula IV should be



LOUIS F. FIESER.

**H. S. Levenson and Hilton A. Smith.** The Saponification of Ethyl Esters of Aliphatic Acids.

Page 1557. Fig. 1, first abscissa legend, read "3050" for "3030."—HILTON A. SMITH.

**H. J. Lucas and W. T. Stewart.** The Oxidation of Alginic Acid by Periodic Acid.

Page 1794. The data regarding the optical rotation of brucine *meso*-tartrate should read "1.2 dm." instead of "1.1 dm."—W. T. STEWART.

**E. P. Linton.** The Dipole Moments of Amine Oxides.

Page 1945. Column 1, line 3 from bottom, read "b. p. 101.30°" for "b. p. 103.30°."—E. P. LINTON.

**Roger Adams and Marvin H. Gold.** Spectra of Certain Dihydroisobenzofurans and Isobenzofurans.

Page 2041. Column 2, lines 12 and 13 under heading "1,3-Dixenylisobenzofuran," omit the words "which turned to a white solid on cooling."—ROGER ADAMS.

**Gilbert N. Lewis and Glenn T. Seaborg.** The Acidity of Aromatic Nitro Compounds toward Amines. The Effect of Double Chelation.

Page 2124. Our attention has been called to an inconsistency. We state that trinitromesitylene showed no color with any base, although in our Table I we show it giving color with ammonia and mono-alkylamine. Mention of the fact that at low temperature it does produce a faint color with these two bases was inadvertently omitted. This behavior in no way contradicts any of our conclusions.—G. N. LEWIS and G. T. SEABORG.

**J. F. Oneto and E. L. Way.** Sulfophenylarsonic Acids and Certain of their Derivatives. III. *p*-Sulfo- and *p*-Sulfonamidodiphenylarsonic Acids.

Pages 2157, 2158. For "*p*-sulfo-, *p*-sulfonamido- and *p*-(*N*-chloro)-sulfonamidodiphenylarsonic acids" read "*p*-sulfo-, *p*-sulfonamido- and *p*-(*N*-chloro)-sulfonamidodiphenylarsinic acids."

Page 2158. In column 2, line 25, for "*p*-sulfonamidotetraphenylarsyl oxide" read "*p,p*'-disulfonamidotetraphenylarsyl oxide."—J. F. ONETO.

**Eugene Pacsu and S. M. Trister.** The Acetone Derivatives of the Mercaptals of Some Monosaccharides. V. The 5,6-Monoacetone Derivative of *d*-Galactosedibenzylmercaptal and the 6-Methyl Derivative of *d*-Galactose.

Page 2304. In column 1, line 9 from the bottom, for "117.5°" read "177.5°."—EUGENE PACSU.

**H. S. Levenson and Hilton A. Smith.** Kinetics of the Saponification of the Ethyl Esters of Several Phenyl Substituted Aliphatic Acids.

Page 2327. Column 2, line 10, read "+I" for "-I."—HILTON A. SMITH.

**Hilton A. Smith and H. S. Levenson.** Kinetics of the Esterification of Cyclohexanoic Acid and of the Saponification of its Ethyl Ester.

Page 2733. The compound referred to as "cyclohexanoic acid" is more properly designated as "cyclohexane-carboxylic acid."—HILTON A. SMITH.

**Paul D. Bartlett and G. Forrest Woods.** Some Reactions of  $\Delta^2$ -Cyclohexenone, Including the Synthesis of Bicyclo(2,2,2)octanedione-2,6.

Page 2934. Formula IX. Delete the question mark after dec.—PAUL D. BARTLETT.

**Edward A. Kelso with W. A. Felsing.** The Pressure-Volume-Temperature Relations of *n*-Hexane and of 2-Methylpentane.

Page 3133. In Table I, data for *n*-Hexane, at 225°C., the third value for moles/liter should be 5.441 instead of 4.441.—W. A. FELSING.